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Covalent Bonding Worksheet #3

Answer Key

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Covalent Bonding ... Chapter 8 Covalent
Bonding and Molecular Structure 8-6 Ex
ample: PF₃ Ex CO₂ Step 1: Count
Valence Electrons Count the total

number of valence electrons in the
molecule or ion. Anions have extra
electrons, so add 1 electron for each
negative charge. Cations have a
deficiency of electrons, so

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A covalent bond in which one atom contributes both bonding electrons (structural formula, Single covalent bond, polyatomic ion, bond dissociation

energy, coordinate covalent bond) coordinate covalent bond. a chemical bond in which one one pair of electrons is shared by two bonded atoms.

8.2 The Nature of Covalent Bonding Flashcards | Quizlet

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