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A certain chemical reaction has the following rate law: $\text{rate} = k[\text{NO}_2][\text{Br}_2]$
Which of the following statements necessarily describe(s) the kinetics of this reaction? I. The reaction is second-order. II. The amount of NO_2 consumed is equal to the amount of Br_2 consumed. III. The rate will not be affected by the addition of a compound other ...

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5. Kinetics; A-Level Specimen Paper 1 - Question 4 (part 6 only) 6. Chemical equilibria; AS Specimen Paper 1 - Question 4; AS Specimen Paper 2 - Question 6 (part 5 only) A-Level Specimen Paper 1 - Question 4 (part 4 only) A-Level Specimen Paper 3 - Question 2 (part 1 only) 8. Thermodynamics; A-Level Specimen

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