
True Self False Self

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Ion Practice Set 1.

In a neutral atom the number of protons equals the number of electrons. 2. An atom can NEVER gain or lose protons 3. The number of protons equals the atomic number NAME Ion Practice Set 1.

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Answer: 6.96 amu 9. Iodine is 80% ^{127}I , 17% ^{126}I , and 3% ^{128}I . Calculate the average atomic mass

of iodine. Answer: 126.86 amu 10. The natural abundance for boron isotopes is 19.9% ^{10}B and 80.1% ^{11}B . Calculate boron's atomic mass. Answer: 126.86 amu 11. Hydrogen is 99% ^1H , 0.8% ^2H , and 0.2% ^3H . Calculate its average atomic mass. Answer: 1.21 amu 12.

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Electron configurations (practice) | Khan Academy

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1 They form a copper(II) complex ion with chloride ions that has a square planar shape. 2 They can be reduced to copper(I) by iodide ions. 23 2They have the electron configuration of $1s^2 2s^2 2p^6 3s^2 3p^6 3d^9$. A 1, 2 and 3 B Only 1 and 2 C Only 2 and 3 D Only 1 Your answer [1]

Practice paper - Set 2 - The Exam Formula

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