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Calculating Ph Pogil Chemistry Answers

Calculating Ph Pogil Answer Key pH =
14.000 - pOH = 14.000 - 2.187 = 11.813
4) -A solution is created by measuring
3.60 x 10³ moles of NaOH and 5.95 x 10⁻⁴
moles of HCl into a container and then
water is added until the final volume is
1.00 L. Calculating pH and pOH
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Public Schools pOH = -log[OH-] = -
log(6.2 x 10⁻⁵) = 4.21 pH = 14.00 - pOH
= 14.00 - 4.21 = 9.79 6) +A -solution
with a H concentration of 1.00 x 10⁷ M is
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