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1. A mixture (is/is not) a chemical combining of substances. 2. In a compound the (atoms/molecules) are (chemically/physically) combined so that the elements that make up the compound (retain/lose) their identities and (do/do not) take on a new set of properties. 3.

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Supplemental Problems Answer Key Chemistry: Matter and Change 55 b. Ca n 5 4 c. Sn n 5 5 15. Write the electron configuration for each element described below and identify the element. a. an element that contains 8 electrons 1s 2s 2p 4 b. an element that contains 14 electrons 1s 2s 2p 6 3s 2 3p 2 The element is silicon. Chapter 6

ANSWER KEY - HONORS CHEMISTRY

(b) Calculate the value of ΔH° , in kJ mol⁻¹, for the combustion of glucose. 2.50 g $\times$ 612 6 1 mol C H O 180.16 g C H O 612 6 = 0.0139 mol C 6 H 12 O 6 39.0 kJ 0.0139 mol = -2,810 kJ mol⁻¹ 1 point is earned for the correct answer. (c) When oxygen is not available, glucose can be oxidized by fermentation. In that process, ethanol and

AP Chemistry 2011 Scoring Guidelines Form B

Answer Key 1.c 2.a 3.c 4.b 5.d 6.false 7.macroscopic 8.true 9.false 10.true 11.false 12.applied 13.pure 14.c 15. a.the initial research was most likely a pure chemistry type of project, learning how to make polymers and studying the properties of these reactions. b.the applied chemistry came in when they wanted to make a specific product with ...

CK-12 Chemistry

(a) (b) Answer is one of the following. A/yellow: mass = 65.14 kg, volume = 3.38 L, density = 19.3 kg/L, likely identity = gold. B/blue: mass = 0.64 kg, volume = 1.00 L, density = 0.64 kg/L, likely identity = apple.

Answer Key Chapter 1 - Chemistry 2e | OpenStax

Explain your answer. Both the reactant and product contains COOH group, while the reactant has -OH group from alcohol and the product does not. Spectrum A has a broad peak at around 3200 cm⁻¹ indicating the -OH group of alcohol. Spectrum B has a broad peak at 2900 cm⁻¹ indicating the -OH group of acid.

[Answer Key] Analytical Chemistry Exercise

If you need the answer key, you can log into the 1.0 experience (use the same URL you are using now, just remove the ".v2" and follow these steps to print an answer key. Generate answer keys for a blank test using the syllabus. Enter the course Syllabus. Preview the assessment.

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