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The molecular formula of the substance is: **Use atomic masses of C = 12 amu, H = 1 amu, O = 16 amu** A. C 3 H 6 O 3 B. C 4 H 26 O C. C 4 H 10 O 2 D. C 5 H 14 O Question 3 A substance of phosphorus (P) and oxygen (O) is found to have a mole ratio of 0.4 moles of P for every mole of O.

Chemical Formulas Practice Test Questions

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