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and Light Explained
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atom?* **What Is Light?**
**How Small Is An
Atom? Spoiler: Very
Small.** Quantum
Mechanics for
Dummies All photons
travel at the speed of
light

Pure Substances and
Mixtures!
(Classification of
Matter)

How is light emitted

from an atom?

Photons, light \u0026amp; electromagnetic waves
The Periodic Table:
Atomic Radius,
Ionization Energy, and
Electronegativity

Emission Spectra and
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Electron Energy and Light

Electron Energy and Light 1... Flinn Scientific, Inc. and HSPIThe POGIL Project.. (electricity) when they move from

lower energy levels
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reveal the behavior of
electrons in an atom?
Type your answers in
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before completing the

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Course Astronomy #24
1. Select the "Color &
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choose highlighter,
choose the appropriate
color, and trace the
arrows in Model 1 and
shade in the table with
the appropriate color.

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Light 3 11. Consider
the hydrogen spectrum
in Model 2.

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Which of the electron
transitions involves the
most energy? $n=6$ to
 $n=2$, violet light, 410
nm. b. Explain why this
transition involves the
most energy based on
your understanding of
the attractive forces
between the electrons
and protons in the
atom. As the electron
moves toward the
nucleus of the atom

there is a stronger
attraction.

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Light Pogil. STUDY.
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What happens when
white light passes
through a prism. The
light bends as it passes
through the prism and
then the white light
separates into different
colors with different
wavelengths.

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sentences describes
the relationship
between wavelength
and energy of light?
(#6c) answer choices .
As wavelength gets
shorter the energy ...

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