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Practical Medicinal Chemistry Laboratory Manual

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...
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The units of volume commonly used in this manual are cm³ and dm³. The conversion between cm³ and dm³ is 1 dm³ = 1000 cm³. Note that a cm³ are the same as millilitres (ml) and dm³ are the same as litres (L). 1 cm³ = 1 ml and 1 dm³ = 1 L. B. Concentration Concentration is a measure of amount per volume. The amount can be grams, kilograms,

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1) Collect the following items which you will need to carry out the experiment. (a) 2 100mL round bottom flask (b) 1 spatula (c) 1 cork ring (d) 1 plastic bowl (e) 1 ceramic Buchner funnel (f) 1 watch glass (g) 1 magnetic stirrer bar (h) 1 zip lock bag. 2) Put on a pair of gloves before you start the experiment.

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...
Step 1: First step involves the formation of Acetylium ion, the reaction is enhanced by proton donation by concentrated acid.
Step 2: Amino group of Para-Aminophenol dissociates as follows:
Step 3: Acetylium ion attacks the anion at the nitrogen atom to form Para-acetyl Aminophenol or Paracetamol.

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