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The amount of thermal energy supplied;
The mass of the substance; What the substance is; Different substances absorb thermal energy at different rates. The specific heat capacity of a substance is the energy needed to raise the temperature of 1kg of the substance by 1 °C $\Delta E = mc\Delta\theta$ ΔE = energy transferred (J) m = mass (kg)

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HEAT ENERGY UNITS calorie: the amount of heat energy required to raise the temperature of 1 gram of water from 145°C to 155°C (also called the "15-degree calorie" or the "little calorie")
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until equilibrium is reached. 3.3 Describe the relationship between average molecular kinetic energy and temperature. Recognize that energy is absorbed when a substance changes from a solid to a liquid to a gas, and that energy is released ...

Physics Thermal Energy, Heat, and Temperature Vocabulary ...

Oct 09 2020 Physics-Thermal-Energy-Test-Study-Guide-Answers 2/3 PDF Drive - Search and download PDF files for free. Write the letter of the choice that best completes each statement 1 The measure of the hotness of an object is its a entropy c specific heat b heat of

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PHYSICS IGCSE 2012 EXAM

REVISION NOTES

The U.S. engineering unit of heat is the British thermal unit (BTU). It is related to the calorie and the joule: $1 \text{ BTU} = 252 \text{ calories} = 1.054 \text{ kJ}$. These reversible conversions of heat energy and work are called the mechanical equivalent of heat. Heat transfer. The heat energy (Q) transferred into or out of a system is given by $Q = mc\Delta T$. The temperature change is positive for a gain in heat energy and negative for heat removed from the object.

Heat - CliffsNotes Study Guides

Physics Thermal Energy Study Guide Key The rise of the Internet and all technologies related to it have made it a lot easier to share various types of information Unfortunately, sometimes the huge amount of information available online is a curse rather than a blessing: many websites just do not