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Homework ...

ANSWER KEY 1 Modern chemistry answer key chapter 2. a 2. b 3. c 4. c 5. a 6. b 7 Modern chemistry answer key chapter 2. c 8. Polarity in a water molecule is caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or basic.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: 4 a. S in H₂SO₃ 6 b. S in MgSO₄ 2 c. S in K₂S 1 d. Cu in

Cu 2S 6 e. Cr in Na
2CrO 4 5 f. N in HNO 3
4 g. C in (HCO 3 ...

7 Chemical Formulas and Chemical Compounds

CHAPTER 5 REVIEW

The Periodic Law

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

CHAPTER 3 REVIEW

Atoms: The Building

Blocks of Matter

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

3 Atoms: The Building Blocks of Matter

CHAPTER 10 REVIEW

States of Matter

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1.

Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and

nonconducting a
metallic crystal (c)
typically has the lowest
melting point of the
four

10 States of Matter - Ms. Agostine's Chemistry Page

CHAPTER 6 REVIEW
Chemical Bonding
SECTION 3 SHORT
ANSWER Answer the
following questions in
the space provided. 1.
a The notation for
sodium chloride, NaCl,
stands for one (a)
formula unit. (c)
crystal. (b) molecule.
(d) atom. 2. d In a
crystal of an ionic
compound, each cation
is surrounded by a
number of (a)
molecules. (c) dipoles.
(b) ...

6 Chemical Bonding

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CHAPTER 2 REVIEW
Measurements and
Calculations SECTION 1
SHORT ANSWER
Answer the following
questions in the space
provided. 1. Determine
whether each of the
following is an example
of observation and
data, a theory, a
hypothesis, a control,
or a model.
observation and data
a. A research team
records the rainfall in
inches

2 Measurements and Calculations

CHAPTER 4 REVIEW
Arrangement of
Electrons in Atoms
SECTION 3 SHORT
ANSWER Answer the
following questions in
the space provided. 1.

State the Pauli exclusion principle, and use it to explain why electrons in the same orbital must have opposite spin states. The Pauli exclusion principle states that no two electrons in an atom may have the

4 Arrangement of Electrons in Atoms

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**Kristy Blankenship -
Boyd County High
School (9-12)**

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this set (26)
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radiation. A form of
energy that exhibits
wavelike behavior as it
travels through space
(3.00×10^8 m/s) ...

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