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### **Chemistry Chapter 10 Energy**

Thermodynamic - study of energy . First law: energy of the universe is constant because it flows between systems (this can be interchanged with the law of conservation of energy) Internal energy:  $E=q+w$   $q=mc(T_{\text{final}}-T_{\text{initial}})$   $c=q/mT$  Energy Units: Joules (J) Calories (Cal) 1 cal = 4.184 J; Enthalpy - flow of energy .

Exothermic -  $-\Delta H$ ; Endothermic -  $+\Delta H$

### **Chapter 10 - Energy - Chemistry**

the ABILITY to do work or produce heat. Ex. energy released when gasoline is burned has the ABILITY to move a car or (do work).

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